# SYNTHESIS OF SOME 3-ALKOXYCARBONYL-3-C-CYANO-3-DEOXY-GLYCOSIDES BY THE REACTION OF 1,5-DIALDEHYDES WITH CYANOESTERS

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### ABSTRACT

The reaction of  $\alpha$ -(R)-methoxydiglycolaldehyde (1) with ethyl and tert-butyl cyanoacetate yielded methyl 2,4-di-O-acetyl-3-C-cyano-3-deoxy-3-ethoxycarbonyl- $\beta$ -D-xylo- (4) and - $\alpha$ -L-xylo-pentopyranosides (5) (isolated as acetyl derivatives), and methyl 3-tert-butoxycarbonyl-3-C-cyano-3-deoxy- $\beta$ -D-xylo- (6) and - $\alpha$ -L-xylopentopyranosides (7), respectively (1:1 addition products) (major products). The minor 1:2 addition products methyl 2-O-acetyl-3-tert-butoxycarbonyl-4-(1-tertbutoxycarbonyl-1-cyanomethyl)-3-C-cyano-3,4-dideoxy- $\alpha$ -L-lyxo- (12) and - $\alpha$ -Lxylo-pentopyranosides (13), and methyl 4-O-acetyl-3-tert-butoxycarbonyl-2-(1-tertbutoxycarbonyl-1-cyanomethyl)-3-C-cyano-2,3-dideoxy-B-D-xylo-pentopyranoside (14) were also isolated in the reaction of 1 with tert-butyl cyanoacetate followed by acetylation. Isomerizations were observed in the acetylation of 6 and 7, yielding methyl 2,4-di-O-acetyl-3-tert-butoxycarbonyl-3-C-cyano-3-deoxy-β-D-xylo- (8) and  $-\alpha$ -L-arabino- (9), and  $-\alpha$ -L-lyxo- (10) and  $-\alpha$ -L-xylo-pentopyranosides (11), respectively. Methyl 2,4,6-tri-O-acetyl-3-C-cyano-3-deoxy-3-ethoxycarbonyl-α-D-gluco-(15),  $-\alpha$ -D-manno- (16), and  $-\beta$ -L-gluco-hexopyranosides (17) together with 3,4-dideoxyhexopyranosides (18 and 19) (1:2 addition products) were isolated in the reaction of  $\alpha$ -(S)-methoxy- $\alpha'$ -(R)-hydroxymethyldiglycolaldehyde (2) with ethyl cyanoacetate after acetylation. The same reaction using  $\alpha$ -(R)-methoxy- $\alpha'$ -(R)hydroxymethyldiglycolaldehyde (3) gave methyl 2,4,6-tri-O-acetyl-3-C-cyano-3deoxy-3-ethoxycarbonyl-β-D-gluco-hexopyranoside (20) after acetylation. Reaction of 3 with tert-butyl cyanoacetate led to methyl 2,6-di-O-acetyl-3-tert-butoxycarbonyl-4-(1-tert-butoxycarbonyl-1-cyanomethyl)-3-C-cyano-3,4-dideoxy-β-D-gluco-hexopyranoside (21; 1:2 addition product) and methyl 2,4,6-tri-O-acetyl-3-tert-butoxycarbonyl-3-C-cyano-3-deoxy-β-D-gluco-hexopyranoside (22; 1:1 addition product), after acetylation.

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# INTRODUCTION

The present work is part of a programme on the synthesis of 3-deoxy-C-glycosyl derivatives and 3-deoxyglycosides, branched at C-3, by the reaction of 1,5-dialdehydes with active methylene compounds.

We have reported¹ on the reaction of thioglycolaldehyde, diglycolaldehyde,  $\alpha$ -(S)-(3-ethoxycarbonyl-2-methylfur-5-yl)diglycolaldehyde,  $\alpha$ -(S)-(3-acetyl-2-methylfur-5-yl)diglycolaldehyde,  $\alpha$ -(R)-methoxydiglycolaldehyde (1), and  $\alpha$ -(S)-methoxy- $\alpha$ '-(R)-hydroxymethyldiglycolaldehyde (2) variously with 2,4-pentanedione, ethyl cyanoacetate, malononitrile, cyanoacetamide, and ethyl and *tert*-butyl cyanoacetate. We now report the reactions of ethyl and *tert*-butyl cyanoacetate with 1, 2, and  $\alpha$ -(R)-methoxy- $\alpha$ '-(R)-hydroxymethyldiglycolaldehyde (3).

### RESULTS AND DISCUSSION

The reactions were carried out in aqueous 1,4-dioxane at room temperature, using piperidine (1%) as the catalyst and a 1:1 molar ratio of dialdehyde and active methylene compound. 1,5-Dialdehydes exist in equilibrium with the cyclic hydrated form<sup>2</sup> but, for simplicity, **1-3** have been depicted as dialdehydes.

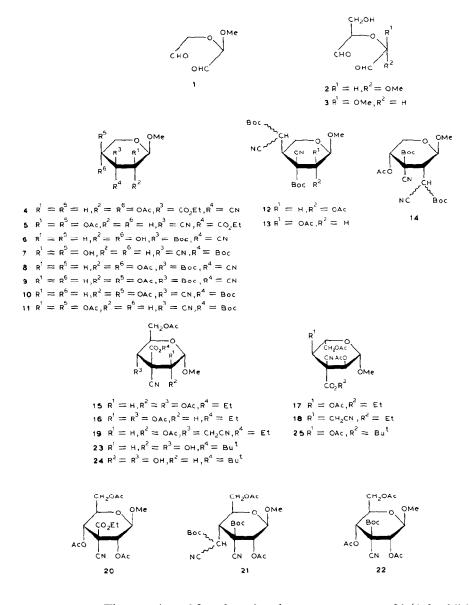
The products were isolated after acetylation of the crude product mixture (4, 5, and 20) or after acetylation of the individual components isolated by column chromatography (8-19, 21, 22, and 25).

The reaction of 1 with ethyl cyanoacetate gave the diacetates 4 (minor) and 5 (major). When 1 reacted with *tert*-butyl cyanoacetate for 2.5 h, 6 and 7 were obtained, corresponding to 1:1 addition products, together with a small amount of a 1:2 addition product (12). The same reagents and longer reaction time (4 h) gave also the major products 6 and 7, and two minor products (13 and 14) of 1:2 addition.

The reaction of **2** with ethyl cyanoacetate for 20 h gave a 4:1 mixture ( ${}^{1}$ Hn.m.r. data) of the  $\alpha$ -D isomers **15** and **16**\* and the  $\beta$ -L isomer **17**. When a longer reaction time (84 h) was used, a 10:1 mixture of **15** and **16** was obtained together with the 1:2 addition and de-ethoxycarbonylation products **18** and **19** (see Scheme 1). Epimerization involving the carbon atom bearing the hydroxymethyl group occurred in the formation of **17** and **18**. This effect was not observed previously in the reaction of **2** with *tert*-butyl cyanoacetate<sup>1g</sup>. When this reaction was re-investigated using the same conditions<sup>1g</sup> (96-h reaction time), **25** was isolated together with the reported compounds **23** + **24**. A similar epimerization has been reported<sup>3</sup> in the cyclization of **2** and **3** with nitroethane. The fact that this isomerization also occurs under the less basic conditions used here supports the view that it is unlikely to occur in the cyclized product.

Compound 20 was the only product isolated after the reaction of 3 with ethyl

<sup>\*</sup>This mixture showed only one spot in t.l.c. and could not be resolved.



cyanoacetate. The reaction of 3 and tert-butyl cyanoacetate gave 21 (1:2 addition product) and 22 (1:1 addition product).

Conventional treatment of 6 and 7 with acetic anhydride-pyridine yielded 8 and 9, and 10 and 11, respectively. This fact could be explained by a ring opening-closure process. Similar isomerizations were observed in acetylations of some 3,3-diacetylpentopyranosides obtained in the reaction of 1 with 2,4-pentanedione. These observations support also the reversibility of this kind of reaction 1d.

Scheme 1. Reaction of 1 with ethyl cyanoacetate (depicted only for 19).

**IABLE I** 

<sup>1</sup>H-n.m.r. CHEMICAL SHIFT DATA FOR 4-22 AND 25

Compound	H-J	H-2	H-4	H-5e	H-5a	9-H	,9-H	Others
<b>4</b> a,d	4.55d	5.15d	5.32dd	4.10m	3.65dd			4.20 (q, 2 H, J7.0 Hz, CH, CH, CO), 3.42 (s, 3 H, MeO), 2.10,
<b>5</b> a.d	4.90d	5.20d	5.25dd	4.00-3.6 0m	0 <b>m</b>			2.06 (2 s, 6 H, 2 Ac), and 1.25 (t, 3 H, 77.0 Hz, CH, CH, O) 4.25 (q, 2 H, 77.0 Hz, CH, CH <sub>2</sub> O), 3.45 (s, 3 H, MeO), 2.12,
<i>p</i> , <i>q</i> <b>9</b>	4.37d	3.66dd/	4.05m8	3.81dd	3.55dd			2.10 (25, 6 H, 2 Ac), and 1.25 (1, 3 H, $J$ / 0 Hz, $CH_3CH_2O$ ) 5.21 (d. 1 H, $J$ 4 T Hz, HO-C2), 5.16 (d. 1 H, $J$ 5.8 Hz,
Tb.e	4.65bd	3.94dd <sup>#</sup>	4.05m <sup>i</sup>	3.51dd	3.69d			HO-C4), 3.45 (s, 3 H, MeO), and 1.48 (s, 9 H, Me <sub>3</sub> C) 5.17 (d, 1 H, 15.8 Hz, HO-C4), 4.58 (d, 1 H, 19.4 Hz,
<b>9</b> a,d	4.60d	5.25d	5.37dd	4.10dd	pseudo-t			HO-C2), 3.38 (s, 3 H, MeO), and 1.48 (s, 9 H, Me <sub>3</sub> C), 3.47 (s, 3 H, MeO), 2.15, 2.10 (2 s, 6 H, 2 Ac),
<b>9</b> a.d	4.54d	5.55d	5.40pt	4.0	4.0 5m			and 1.41 (8, 9 H, Me <sub>3</sub> C) 5.08 (8, 3 H, Me0), 2.15, 2.10 (2 s, 6 H, 2 Ac),
10a,d	4.60d	5.40d	5.55dd	3.7	3.7 5m			3.45 (s, 3 H, MeO), 2.15, 2.10 (2 s, 6 H, 2 Ac),
11a.d	4.85d	5.17d	5.27dd	4.00-3.6 0m	<b>m</b> 0			and 1.4 (s, 9 H, Me <sub>3</sub> C) 3.45 (s, 3 H, MeO), 2.12, 2.10 (2 s, 6 H, 2 Ac),
12c.d	4.62d	5.30d	3.20ddd	3.80dd	4.10			and 1.45 (8, 9 H, Me <sub>3</sub> C) and 1.42 (4, 1 H, 12.3 Hz, NC-CH-Boc), 3.45 (8, 3 H, MeO), 2.06
13ª.d	4.90d	5.20d	3.10-2.90m	3.80dd	pseudo-t 4.16			(s, 3 H, MeCOU), 1.35, and 1.44 (2s, 18 H, 2 Mes C) 3.62 (d, 1 H, J.3.2 Hz, NC-CH-Boc), 3.45 (s, 3 H, MeO), 2.15
<b>14</b> a,d	4.80d	3.05dd	5.35dd	4.12dd	3.70dd			(s, 3 H, MeCOO), 1.32, and 1.30 (2 s, 18 H, 2 Mes.C) 3.55 (s, 3 H, MeO), 3.45 (d, J 1.5 Hz, NC-CH-Boc), 2.12
15 <sup>b,d</sup> 16 <sup>b,d</sup> 17 and 20 <sup>p,d</sup>	4.90d 4.57d 4.70d	5.17d 5.30d 5.35d	5.13d 5.45d 5.40d		4.10m	4.24dd 4.15-4.00m 4.35-3.90m	4.00m	(\$, 3 H, MeC.O.U, and 1.30 (\$, 18 H, 2 Me <sub>3</sub> C.) 4.10 (m, 2 H, CH <sub>2</sub> CH <sub>2</sub> O) 3.31 (\$, 3 H, MeO), 2.05-1.90 (5 s, 9 H, 3 AeO, 1.126 and 1.24 (2 t, 3 H, CH <sub>3</sub> CH <sub>2</sub> O) 4 3 45-3 90 (m, 2 H, CH <sub>2</sub> CH <sub>2</sub> O) 3 50 (\$, 3 H, MeO), 2 1.2 2 10
<b>18</b> a.a	4.65d	5.14d	3.00-2.75m		3.90dt	4.50-4.2 0m	2 0m	(2s, 9 H, 3 Ac), and 1.25 (t, 1 H, 77.0 Hz, CH,CH,O) 4.50-4.20 (m, 2 H, CH,CH,O), 3.50 (s, 3 H, MeO), 2.65-2.50 (m, 2 H, CH,CN), 2.15 (2s, 6 H, 2 MeCOO), and 1.35
p.q <b>61</b>	5.00d	5.10d	2.90-2.75ш		4.05 pseudo-t	4.45-4.2 5m	2 5m	(t, 3 H, J 7.0 Hz, CH <sub>3</sub> CH <sub>2</sub> O) 4.45-4.25 (m, CH <sub>3</sub> CH <sub>2</sub> O), 3.45 (s, 3 H, MeO), 2.57 (m, 2 H, CH <sub>2</sub> CN), 2.16, 2.15 (2s, 6 H, 2 MeCOO), 1.36 (t, 3 H,
<b>21</b> c,d	4.62d	5.27d	3.18dd		4.23-4.1 7m	1 7m	4.32dd	J 1.3 H2, CH <sub>3</sub> -CH <sub>3</sub> OJ 3.48 (s, 3 H, MeO), 3.47 (d, 1 H, J 0.8 Hz, NC-CH-Boc), 2.08, 2.06 (2 s, 6 H, 2 MeCOO), 1.48, and 1.44 (2 s, 18 H,
<b>22</b> and <b>25</b> °,4 4.57d	4.57d	5.25d	5.33d		3.94ddd	4.23dd	4.04dd	2. Me3.7.) 3.43 (3.41, MeO), 2.03, 2.01, 1.99 (3.8, 9 H, 3 Ac), and 1.34 (s. 9 H. Me.C)

\*80 MHz. \*200 MHz. \*500 MHz. \*fror solutions in CDCl<sub>3</sub> (internal Me<sub>3</sub>Si). \*fror solutions in (CD<sub>3</sub>)<sub>2</sub>CO. /d. J. 7.7 Hz, after isotopic change. \*6d. J. 10.3 Hz, after isotopic change. \*hd. J. 3.8 Hz, after isotopic change. \*ldd. J. 11.0 and 5.0 Hz, after isotopic change. \*pseudo-t, after decoupling in the signal corresponding to H-1.

10

11

12

13

14

15

16

18

19

21

17 and 20

22 and 25

1.5

3.5

1.5

3.5

8.5

3.6

1.5

8.0

8.0

3.5

8.1

8.1

7.0

6.0

4.6

4.5

5.0

9.0

10.0

11.8

11.5

10.0

10.0

0.01

9.7

10.0

0.01

9.8

10.0

4.5

6.0

5.7

4.3

12.0

13.6

12.3

 ${}^{1}C_{4}(L)$ 

 $C_{4}(L)$ 

 ${}^{1}C_{4}(1)$ 

 ${}^{1}C_{4}(L)$ 

 ${}^4C_1(D)$ 

 ${}^4C_1(D)$ 

 ${}^4C_1(\mathbf{D})$ 

 ${}^{1}C_{4}(L)$ 

 ${}^4C_1(D)$ 

 ${}^4C_1(D)$ 

 ${}^{1}C_{4}(L)$  and  ${}^{4}C_{1}(D)$ 

 ${}^4C_1(D)$  and  ${}^1C_4(L)$ 

Compound	J <sub>1.2</sub>	J <sub>4,5e</sub>	J <sub>4,5a</sub>	J <sub>5e,5a</sub>	J <sub>5.6</sub>	J <sub>5,6'</sub>	J <sub>6,6</sub>	Conformation
4	7.0	4.7	8.9	12.0				${}^{4}C_{1}(D)$
5	3.4	5.8	9.7					${}^{\mathrm{I}}C_{\mathtt{t}}(\mathtt{L})$
6	7.7	4.7	10.3	11.7				${}^4C_1(D)$
<b>7</b> <sup>a</sup>	3.8	5.0	11.0	11.2				${}^{1}C_{4}(L)$
8	7.0	4.4	8.8	11.7				${}^4C_1(D)$
9	7.5	ь	ħ					4C(1)

TABLE II

11.8

11.5

12.0

The structures of 4-22 and 25 were established on the basis of elemental analysis and spectroscopic data (Tables I-III).

2.7

6.0

2.6

The configurations at C-2 and C-4, and the preferred conformations were deduced from the values of  $J_{1,2} J_{4,5a}$ , and  $J_{4,5e}$  in the <sup>1</sup>H-n.m.r. spectra. Compounds **4, 6, 8, 9, 14, 17, 18, 20–22,** and **25** showed  $J_{1,2}$  values of 8.5–7.0 Hz, indicating H-1,2 to be trans-diaxial. The same inference applies to H-4,5a in 4-8, 10-22, and **25**  $(J_{4.5a}$  11.8–8.8 Hz). The  $J_{4.5a} + J_{4.5e}$  value of 4.0 Hz for **9** reflected a synclinal relationship between H-4 and H-5a,5e, the  $J_1$ , values of 3.4–3.8 Hz for 5, 7, 11, 13, **15,** and **19** indicated H-1,2 to be equatorial, axial, and the  $J_{1,2}$  values of 1.5 Hz for 10, 12, and 16 indicated H-1,2 to be equatorial, equatorial.

The configuration at C-3 in each compound was assigned tentatively on the basis of the expected higher stability of equatorial alkoxycarbonyl and axial cyano groups. Supporting this assignment is the trans-elimination4 of some 3-tertbutoxycarbonyl-3-C-cyano-3-deoxy-C-glycosyl and 3-tert-butoxycarbonyl-3-Ccyano-3-deoxyglycoside derivatives to yield 3-cyano- $\Delta^2$ - and - $\Delta^3$ -dihydropyran derivatives.

Table III includes the <sup>13</sup>C-n.m.r. data for 4-22 and 25. The resonances of C-1 in 4, 6, 8, 9, 14, 18, 20-22, and 25 (103.0-100.0 p.p.m.) and 5, 7, 10, 11, 12, 13, 15, 16, and 19 (97.10-94.90 p.p.m.) accorded<sup>5-7</sup> with an equatorial and axial disposition respectively, of the methoxyl group.

On the basis of  $J_{1,2}$  and  $J_{4,5}$  values (see Table II), an isomerization at C-5 and a preferred  ${}^{1}C_{4}(L)$  conformation is assumed for 17, 18, and 25. This inference is

 $<sup>^{</sup>a4}J_{1,5a} \sim 0.6 \text{ Hz.}$   $^{b}J_{4,5} + J_{4,5'} \sim 4.0 \text{ Hz.}$   $^{c}J_{5,6} + J_{5,6'} \sim 6.0 \text{ Hz.}$ 

TABLE III

3C-N.M.R. CF	TEMICAL SE	$^{13}\mathrm{C}$ -n m.r. chemical shift data for <b>4–22</b> and <b>25</b>	2 AND 25				ļ	
Compound	C-1	C-2,C-4	C-5	C-6	C-3	МеО	CN	Others
<b>4</b> a, d	100.05	69.06, 67.81	62.18		54.10	56.73	113.63	168.91, 168.62, 163.62 (3 COO), 64.07 (OCH <sub>2</sub> CH <sub>3</sub> ),
Sa, d	94.90	69.54, 67.92	56.20		52.15	55.62	113.64	169.14, $168.86$ , $169.14$ , $168.86$ , $169.14$ , $169.14$ , $169.14$ , $169.14$ , $169.14$ , $169.14$ , $169.14$ , $169.14$ , $169.14$ , $199.$
o.60.e	103.04	71.58, 69.21	90.99		60.65	56.24	115.52	20.25 (McCOO), and 13.70 (CCH <sub>2</sub> CH <sub>3</sub> ) 165.48 (COO), 83.66 (CMe), and 27.31 (CMe <sub>3</sub> )
76.€	20.06	70.17, 68.14	59.33		0	55.18	0	84.04 (CMe <sub>3</sub> ), and 27.35 (CMe <sub>3</sub> )
<b>8</b> a.d	100.16	69.02, 67.84	62.26		55.60	56.25	113.93	168.85, 168.45, 161.98 (3 COO), 85.94 (CMe <sub>3</sub> ),
<b>9</b> 0.d	101.04	70.36, 66.73	63.74		52.38	56.82	114.08	27.46 (CMe <sub>3</sub> ), and 20.41 (MeCOO) 169.14, 168.37, 161.16 (3 COO), 85.68 (CMe <sub>3</sub> ),
10a.d	96.84	68.96, 67.79	62.20		48.80	56.75	113.78	27.58 (CMe <sub>3</sub> ), and 20.67 (MeCOO) 169.09, 168.45, 161.88 (3 COO), 85.21 (CMe <sub>3</sub> ),
11a,d	95.02	69.72, 68.01	56.41		53.22	55.78	114.15	27.41 (CMe <sub>3</sub> ), and 20.37 (MeCOO) 169.33, 168.97, 163.17 (3 COO), 85.78 (CMe <sub>4</sub> ),
12c.d	96.93	96.89	57.07		46.19	55.56	113.66	27.56 (CMe <sub>3</sub> ), and 20.51 (MeCOO) 168.60, 163.20 (3 COO), 86.10, 85.40 (2 CMe <sub>3</sub> ).
7 7 7	000	20 30	00 /4				113.62	27.83, 27.65 (2 CMe <sub>3</sub> ), and 20.65 (MeCOO)
1.5a.	87.56	/0.208	20.88		21.05	22.82	112.94	169.32, 163.8, 162.68 (3 COO), 86.58, 85.83 (2 CMe <sub>3</sub> ), 27.75, 27.67 (2 CMe <sub>3</sub> ), and 20 60 (MeCOO)
<b>14</b> a,d	101.0	09:89 4	62.83		54.15	57.30	113.26	168.74, 162.62 (2 COO), 86.61 (2 CMe <sub>3</sub> ),
15b.d	95.24	69.23, 67.02	65.15	61.29	52.28	55.59	113.54	27.55, 27.53 (2 CMe <sub>3</sub> ), and 20.37 (2 MeCOO) 170.17, 168.90, 168.53, 164.13 (4 COO), 64.04 (OCH, CH <sub>3</sub> ).
$16^{b,d}$	97.10		0		19.40	55.11	o	20.66, 20.17 (2 MeCOO), and 13.55 (OCH, CH,)
17 and 20ª,d	100.72	72.29, 69.22	15.79	61.64	56.33	57.15	113.61	170.44, 168.41, 168.23, 163.28 (4 COO), 64.24 (OCH <sub>2</sub> CH <sub>3</sub> ),
<b>18</b> <i>a,d</i>	100.35	i 39.06		j	55.26	57.00	115.70	(20.03, 20.34 (3 MeCOO), and 13.74 (OCH <sub>2</sub> CH <sub>3</sub> ) 170.40, 168.60 (2 COOMe), 164.24 (COOEt), 20.70, 20.35
10a.d	95.72	k 29.24	×	-	51.28	55.8%	113.70	(2 MeCOO), 15.91 (CH <sub>2</sub> CN), and 13.66 (OCH <sub>2</sub> CH <sub>3</sub> ) 170 40 168 9072 COOMe) 165 50 (COOFt) 20 60 20 10
ì							114.11	(2 MeCOO), 16.10 (CH <sub>2</sub> CN), and 14.10 (OCH <sub>3</sub> CH <sub>3</sub> )
<b>21</b> c. <i>d</i>	100.41	E E	ŧ	63.03	56.10	57.11	113.39 112.93	170.22, 168.43 (2 COOMe), 162.92, 162.16 (2 COOMe <sub>3</sub> ), 87.10, 86.26 (2 CMe <sub>3</sub> ), 27.63, 27.46 (2 CMe <sub>3</sub> ), 20.85, and
22 and 25c.d	100.57	72.14, 69.10	67.24	61.45	57.36	57.06	113.79	20.50 (2 MeCOO) 170.36, 168.15, 168.04, 161.61 (4 COO), 86.14 (CMe <sub>3</sub> ), 27.19 (CMe <sub>3</sub> ), 20.54, 20.28, and 20.18 (3 MeCOO)

\*20 MHz. \*50 MHz. \*75 MHz. \*For solution in CDCl<sub>3</sub>. \*For solution in (CD<sub>2</sub>), CO. Signals at 37.30 and 32.36 correspond to C-4, NC-CH-Boc. \*Signals at 43.44 and 37.38 correspond to C-2, NC-CH-Boc. Signals at 71.95 and 63.00 correspond to C-4, NC-CH-Boc. \*Signals at 71.95 and 69.88 correspond to C-5, Signals at 71.95 and 69.88 correspond to C-2, C-5. Signals at 64.23 and 63.25 correspond to C-6, -OCH<sub>2</sub>CH<sub>3</sub>. \*Signals at 71.95 and 69.88 correspond to C-4, NC-CH-Boc. \*Accurate data could not be obtained.

supported by the chemical shift values (100.5  $\pm$ 0.2 p.p.m.) of the resonances for C-1 in these compounds, which accord with an equatorial disposition of the methoxyl group.

In addition to the signals noted above, the 1:2 addition compounds 12–14 and 21 gave <sup>13</sup>C signals for CN (~113 p.p.m.) and CMe<sub>3</sub> groups (87.1–85.4 and 27.8–27.5 p.p.m.). The position of the CN–CH–Boc group was deduced from the chemical shifts of the resonances for H-2,4 and C-2,4 (see Tables I and III). However, 18 and 19 gave signals for two CN groups, but only one each for COOEt and -CH<sub>2</sub>CN groups, suggesting that 18 and 19 had been formed by loss of the COOEt group of the 1-ethoxycarbonyl-1-cyanomethyl moiety linked at C-4 in an initial 1:2 addition product 26, as shown in Scheme 1. The isolation of a compound similar to the intermediate 28 in the reaction of 2 with 2,4-pentanedione<sup>1g</sup> supports the proposed mechanism.

In each of the preferred conformers (see Table II), the bulky substituents are generally equatorial. Only **9**, **10**, **12**, and **16** have an axial AcO group at C-2. These observations accord with thermodynamic control in a reversible process.

Thus, the behaviour of the dialdehyde 1 was similar to that of ethyl and tert-butyl cyanoacetate. However, 2 and 3 gave higher yields of products with tert-butyl cyanoacetate than with ethyl cyanoacetate. 1:1 Addition products with gluco and manno configurations were obtained from 2 and ethyl or tert-butyl cyanoacetate. However, 3 gave only gluco products, reflecting the  $\Delta^2$  effect<sup>8</sup> in manno isomers. Epimerization at the carbon bearing the hydroxymethyl group was observed only in the reactions of 2.

## EXPERIMENTAL

General methods. — Organic solutions were dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Solvents were evaporated under diminished pressure at <40°. Column chromatography was carried out on silica gel (Merck, 70–230 mesh, ASTM). Melting points (uncorrected) were obtained with an Electrothermal apparatus. Optical rotations were measured with a Perkin-Elmer 141 automatic polarimeter and i.r. spectra with a Perkin-Elmer 983 G spectrometer. N.m.r. spectra (internal Me<sub>4</sub>Si) were obtained with a Bruker WP 80, WP 200, or AM 300 spectrometer.

Reactions of 1-3 with ethyl and tert-butyl cyanoacetate. — Ethyl or tert-butyl cyanoacetate and piperidine (0.2 mL) were added to a solution of the dialdehyde (1-3) in aqueous 1,4-dioxane (2:1, 30 mL). Each mixture was stored at room temperature and then concentrated. Water ( $\sim$ 20 mL) was added to the residue, and the mixture was extracted with ethyl acetate (4 × 50 mL). The combined extracts were dried, filtered, and concentrated to give the crude product.

The following amounts and conditions were used.

(a) Reaction of ethyl cyanoacetate with 1. The crude product was treated at room temperature for 16 h with acetic acid-acetic anhydride-acetyl chloride (8:4:16 mL). Column chromatography (3:1 hexane-ether) of the products

Starting compound (g)	Active methylene compound <sup>a</sup> (g)	Time (h)	Products (g, %)
$1^b$	A (1.7)	0.5	4 (0.96, 19.6)
			<b>5</b> (2.72, 27.7)
<b>1</b> <sup>b</sup>	B (2.1)	2.5	6(1.10, 27.2)
			7 (0.53, 14.3)
			<b>12</b> (0.08, 1.2)
<b>1</b> <sup>b</sup>	B(2.1)	4.0	6 (0.46, 11.3)
			7 (0.27, 6.6)
			<b>13</b> (0.30, 4.6)
			<b>14</b> (0.14, 2.2)
<b>2</b> <sup>c</sup>	A (1.7)	20.0	$15 + 16(0.30, 4.5)^d$
	, ,		<b>17</b> (0.30, 4.2)
<b>2</b> <sup>c</sup>	A (1.7)	84.0	$15 + 16(0.30, 4.5)^{e}$
	, ,		<b>18</b> (0.13, 2.0)
			<b>19</b> (0.10, 1.5)
<b>2</b> <sup>c</sup>	B (2.1)	96.0	$23 + 24(1.9, 42.5)^{dlg}$
	` '		<b>25</b> (0.50, 8.0)
<b>3</b> f	A (1.7)	14.0	20 (2.25, 37.4)
<b>3</b> f	$\mathbf{B}(2.1)$	14.0	<b>21</b> (1.18, 18.3)
	` '		<b>22</b> (1.10, 17.1)

<sup>&</sup>lt;sup>a</sup>A, ethyl cyanoacetate; B, tert-butyl cyanoacetate. <sup>b</sup>Prepared from methyl β-D-xylopyranoside (2.44 g, 15 mmol)<sup>1d</sup>. <sup>c</sup>Prepared from methyl α-D-glucopyranoside (2.91 g, 15 mmol)<sup>1e</sup>. <sup>d</sup>Molar ratio ~4:1. <sup>e</sup>Molar ratio ~10:1. <sup>f</sup>Prepared from methyl β-D-glucopyranoside (2.91 g, 15 mmol)<sup>9</sup>.

gave, first, methyl 2,4-di-O-acetyl-3-C-cyano-3-deoxy-3-ethoxycarbonyl- $\beta$ -D-xylo-pentopyranoside (4), m.p. 90° (from hexane),  $[\alpha]_D^{25}$  –56° (c 1, chloroform);  $\nu_{\rm max}^{\rm KBr}$  1759, 1259, 1214, 1073, 1044, 978, 899, and 855 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 51.35; H, 5.50; N, 4.44.  $C_{14}H_{19}NO_8$  calc.: C, 51.06; H, 5.81; N, 4.25%.)

Eluted second was methyl 2,4-di-*O*-acetyl-3-*C*-cyano-3-deoxy-3-ethoxy-carbonyl- $\alpha$ -L-*xylo*-pentopyranoside (**5**), m.p. 79–80° (from hexane–ether), [α]<sub>D</sub><sup>25</sup> +113° (*c* 1, chloroform);  $\nu_{\text{max}}^{\text{KBr}}$  1744, 1285, 1265, 1218, 1140, 1049, 965, and 898 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 50.70; H, 5.82; N, 4.27. C<sub>14</sub>H<sub>19</sub>NO<sub>8</sub> calc.: C, 51.06; H, 5.81; N, 4.25%.)

(b) Reaction of tert-butyl cyanoacetate with 1. Column chromatography (2:1 hexane-ether) of the crude product (2.5-h reaction) gave, first, a product which was treated with acetic acid-acetic anhydride-acetyl chloride (4:2:8 mL). Column chromatography (3:1 hexane-ether) gave only methyl 2-O-acetyl-3-tert-butoxycarbonyl-4-(1-tert-butoxycarbonyl-1-cyanomethyl)-3-C-cyano-3,4-dideoxy- $\alpha$ -L-lyxopentopyranoside (12), m.p. 154-156° (from hexane-ether),  $[\alpha]_D^{24}$  +24° (c 1, chloroform);  $\nu_{max}^{KBr}$  2250, 1748, 1254, 1221, 1143, and 834 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I-III (Found: C, 57.70; N, 7.00; N, 6.54.  $C_{21}H_{30}N_2O_8$  calc.: C, 57.52; H, 6.89; N, 6.38%). A slow-moving complex mixture (0.2 g) was obtained which was not investigated.

Eluted second was methyl 3-tert-butoxycarbonyl-3-C-cyano-3-deoxy-β-D-

*xylo*-pentopyranoside (**6**), m.p. 138–141° (from hexane–ether),  $[\alpha]_D^{25}$  –27° (*c* 1, methanol);  $\nu_{\text{max}}^{\text{KBr}}$  3470, 3380, 2250, 1740, 1250, 1220, 1195, 1140, 1100–1080, 1040, 930, and 838 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 53.03; H, 6.83; N, 5.03. C<sub>12</sub>H<sub>19</sub>NO<sub>6</sub>: calc.: C, 52.73; H, 7.00; N, 5.12%).

Eluted third was methyl 3-tert-butoxycarbonyl-3-C-cyano- $\alpha$ -L-xylo-pentopyranoside (7), m.p. 151–152° (from hexane–ether),  $[\alpha]_D^{25}$  –115° (c 1, methanol);  $\nu_{\rm max}^{\rm KBr}$  3529, 3445, 2258, 1721, 1292, 1065, 1016, 952, and 838 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 53.27; H, 7.12; N, 5.20.  $C_{12}H_{19}NO_6$  calc.: C, 52.73; H, 7.00; N, 5.12%.)

The reaction of **1** with *tert*-butyl cyanoacetate at room temperature (4 h) and column chromatography (2:1 hexane–ether) gave, first, a product that was conventionally acetylated with acetic anhydride–pyridine (6:3 mL). Column chromatography (4:1 hexane–ether) gave methyl 4-*O*-acetyl-3-*tert*-butoxycarbonyl-2-(1-*tert*-butoxycarbonyl-1-cyanomethyl)-3-*C*-cyano-2,3-dideoxy- $\beta$ -D-*xylo*-pentopyranoside (**14**), isolated as a syrup,  $[\alpha]_{4360}^{25}$  +7° (*c* 1, chloroform);  $\nu_{\text{max}}^{\text{film}}$  2253, 1750, 1744, 1280, 1258, 1214, 1096, and 839 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 57.35; H, 6.70; N, 6.65.  $C_{21}H_{30}N_2O_8$  calc.: C, 57.52; H, 6.89; N, 6.38%.)

Eluted second was a product that was conventionally acetylated with acetic anhydride–pyridine (6:3 mL). Column chromatography (2:1 hexane–ether) then gave methyl 2-*O*-acetyl-3-*tert*-butoxycarbonyl-4-(1-*tert*-butoxycarbonyl-1-cyanomethyl)-3-*C*-cyano-3,4-dideoxy- $\alpha$ -L-*xylo*-pentopyranoside (13), m.p. 112–113° (from hexane–ether),  $[\alpha]_D^{20}$  –36° (*c* 1, chloroform);  $\nu_{\text{max}}^{\text{KBr}}$  1740, 1299, 1224, 1160, 1140, 1058, 903, and 833 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 57.35; H, 6.70; N, 6.65.  $C_{21}H_{30}N_2O_8$  calc.: C, 57.52; H, 6.89; N, 6.38%.)

Eluted third and fourth were 6 and 7.

(c) Reaction of ethyl cyanoacetate with **2**. Column chromatography (ether) of the crude product (20-h reaction) gave, first, a product that was treated with acetic anhydride–pyridine (7:5 mL) at  $-10^{\circ}$ . Column chromatography (1:1 hexane–ether) then gave methyl 2,4,6-tri-O-acetyl-3-C-cyano-3-deoxy-3-ethoxycarbonyl- $\beta$ -L-gluco-hexopyranoside (**17**), m.p. 114–115° (from hexane–ether),  $[\alpha]_{\rm D}^{25}$  +25.6° (c 1, chloroform);  $\nu_{\rm max}^{\rm KBr}$  2250, 1766, 1375, 1248–1200, 1161, 1101, 1046, and 909 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 50.80, H, 5.36; N, 3.37.  $C_{17}H_{23}NO_{10}$  calc.: C, 50.87; H, 5.77; N, 3.48%.)

Eluted second was a product that was treated with acetic anhydride–pyridine (10:5 mL). Column chromatography (1:1 hexane–ether) gave methyl 2,4,6-tri-O-acetyl-3-C-cyano-3-deoxy-3-ethoxycarbonyl- $\alpha$ -D-gluco- (15) and - $\alpha$ -D-manno-hexopyranosides (16), isolated as a syrup;  $\nu_{\rm max}^{\rm film}$  1751, 1224, and 1052 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 51.00; H, 5.62; N, 3.65.  $C_{17}H_{23}NO_{10}$  calc.: C, 50.87; H, 5.77; N, 3.49%.)

Column chromatography (ether) of the crude product (84-h reaction) gave, first, a product that was treated with acetic anhydride-pyridine (6:3 mL) at  $-10^{\circ}$ .

Column chromatography (1:1 hexane–ether) then gave methyl 2,6-di-O-acetyl-3-C-cyano-4-cyanomethyl-3,4-dideoxy-3-ethoxycarbonyl- $\beta$ -L-gluco-hexopyranoside (18), m.p. 112–113° (from hexane–ether),  $[\alpha]_D^{2.5}$  +64° (c1, chloroform);  $\nu_{\text{max}}^{\text{KBr}}$  2252, 1747, 1449, 1371, 1238, 1054, 894, and 854 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 53.60; H, 5.71; N, 7.32.  $C_{17}H_{22}N_2O_8$  calc.: C, 53.39; H, 5.80; N, 7.32%.)

Eluted second was a product that was treated with acetic anhydride–pyridine (8:3 mL). Column chromatography (1:2 hexane–ether), gave methyl 2,6-di-O-acetyl-3-C-cyanomethyl-3,4-dideoxy-3-ethoxycarbonyl- $\alpha$ -D-gluco-hexopyranoside (19) isolated as a syrup,  $[\alpha]_D^{27}$  +90° (c 1, chloroform);  $\nu_{\text{max}}^{\text{film}}$  2252, 1749, 1243, 1057, 961, and 854 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 53.22; H, 6.12; N, 7.25.  $C_{17}H_{22}N_2O_8$  calc.: C, 53.39; H, 5.80; N, 7.32%.)

Eluted third was a product that was acetylated with acetic anhydride-pyridine (10:5 mL). Column chromatography (1:1 hexane-ether) then gave a mixture of 15 and 16.

(d) Reaction<sup>1g</sup> of tert-butyl cyanoacetate with 2. Column chromatography (ether) of the crude product gave, first, a product that was treated with acetic anhydride-pyridine (8:4 mL) at  $-10^{\circ}$ . Column chromatography (4:1 hexane-ether) then gave a mixture of products. Crystallisation from 6:1 hexane-ether gave methyl 2,4,6-tri-O-acetyl-3-tert-butoxycarbonyl-3-C-cyano-3-deoxy- $\beta$ -L-gluco-hexopyranoside (25), m.p. 135–136°, [ $\alpha$ ] $_{\rm D}^{25}$  +28° (c1, chloroform);  $\nu_{\rm max}^{\rm KBr}$  1758, 1740, 1216, 1157, 1121, 1056, 902, and 835 cm $^{-1}$ . For  $^{1}$ H- and  $^{13}$ C-n.m.r. data, see Tables I-III. (Found: C, 52.78; H, 5.96; N, 3.21.  $C_{19}H_{27}NO_{10}$  calc.: C, 53.14; H, 6.33; N, 3.26%.)

Eluted second was a mixture of methyl 3-tert-butoxycarbonyl-3-C-cyano-3-deoxy- $\alpha$ -D-gluco- (23) and - $\alpha$ -D-manno-hexopyranosides (24)<sup>1g</sup>.

- (e) Reaction of ethyl cyanoacetate with 3. The crude product was acetylated with acetic anhydride-pyridine (10:5 mL) at  $-10^{\circ}$ . Column chromatography (1:1 hexane-ether) then gave methyl 2,4,6-tri-O-acetyl-3-C-cyano-3-deoxy-3-ethoxycarbonyl- $\beta$ -D-gluco-hexopyranoside (20), m.p. 114-115° (from hexane-ether),  $[\alpha]_D^{25}$  -26° (c 1, chloroform). The i.r. and n.m.r. data were identical to those for 17. (Found: C, 50.90; H, 5.52; N, 3.46.  $C_{17}H_{23}NO_{10}$  calc.: C, 50.87; H, 5.77; N, 3.49%.)
- (f) Reaction of tert-butyl cyanoacetate with 3. Column chromatography (1:6 hexane-ether) of the crude product gave, first, a product that was treated with acetic anhydride-acetic acid-acetyl chloride (6:3:12 mL). Column chromatography (1:1 hexane-ether) then gave methyl 2,6-di-O-acetyl-3-tert-butoxycarbonyl-4-(1-tert-butoxycarbonyl-1-cyanomethyl)-3-C-cyano-3,4-dideoxy- $\beta$ -D-gluco-hexopyranoside (21), m.p. 138–140° (from hexane-ether),  $[\alpha]_D^{20}$  –57° (c 1, chloroform);  $\nu_{\text{max}}^{\text{KBr}}$  1746, 1210, 1136, 1065, 899, and 836 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I-III. (Found: C, 56.77; H, 6.82; N, 5.49.  $C_{24}H_{34}N_2O_{10}$  calc.: C, 56.37; H, 6.87; N, 5.46%.)

Eluted second was a complex mixture (0.86 g) that was not investigated.

Eluted third was a product that was treated with acetic anhydride-pyridine (10:5 mL). Column chromatography of the crude product (1:1 hexane-ether) then gave methyl 2,4,6-tri-*O*-acetyl-3-*tert*-butoxycarbonyl-3-*C*-cyano-3-deoxy-β-D-gluco-hexopyranoside (**22**), m.p. 135–136° (from hexane-ether),  $[\alpha]_D^{20} - 27^\circ$  (*c* 1, chloroform);  $\nu_{\text{max}}^{\text{KBr}}$  1757, 1218, 1156, 1052, 902, and 834 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 52.70; H, 6.02; N, 3.57.  $C_{19}H_{27}\text{NO}_{10}$  calc.: C, 53.14; H, 6.33; N, 3.26%.)

Acetylation of 6 and 7. — Conventional treatment of 6 and 7 with acetic anhydride-pyridine at  $-10^{\circ}$  and extraction of the products into chloroform gave the following results.

Starting compound (g)	$Ac_2O$ -pyridine (mL)	Products (g, %)
6 (0.9) 7 (0.32)	10.5 6:3	<b>8</b> (0.75, 63.7), <b>9</b> (0.20, 16.6) <b>10</b> (0.06, 14.3), <b>11</b> (0.08, 19.1)
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(a) Methyl 2,4-di-O-acetyl-3-tert-butoxycarbonyl-3-C-cyano-3-deoxy- $\beta$ -D-xylo- (8) and - $\alpha$ -L-arabino-pentopyranosides (9). Column chromatography (3:1 hexane-ether) of the crude product gave, first, **8**, m.p. 89–90° (from hexane-ether),  $[\alpha]_D^{25}$  -43° (*c* 1, chloroform);  $\nu_{\text{max}}^{\text{KBr}}$  2245, 1757, 1279, 1212, 1155, 1069, 1041, 981, 898, and 834 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 53.56; H, 6.43; N, 3.83.  $C_{16}H_{23}NO_8$  calc.: C, 53.77; H, 6.48; N, 3.90%).

Eluted second was **9**, m.p. 126–128° (from hexane–ether),  $[\alpha]_{4360}^{25}$  +12° (*c* 1, chloroform);  $\nu_{\rm max}^{\rm KBr}$  2253, 1753, 1276, 1240, 1222, 1154, 1106, 902, 833, and 747 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 53.40; H, 6.31; N, 3.99.  $C_{16}H_{23}NO_8$  calc.: C, 53.77; H, 6.48; N, 3.90%.)

(b) Methyl 2,4-di-O-acetyl-3-tert-butoxycarbonyl-3-C-cyano-3-deoxy- $\alpha$ -L-lyxo- (10) and - $\alpha$ -L-xylo-pentopyranosides (11). Column chromatography (3:1 hexane-ether) of the crude product gave, first, 10, isolated as a syrup;  $\nu_{\rm max}^{\rm KBr}$  2252, 1756, 1260, 1218, 1138, 1088, 1047, 1011, 896, and 837 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 53.85, H. 6.60; N, 3.95.  $C_{16}H_{23}NO_8$  calc.: C, 53.77; H, 6.48; N, 3.90%).

Eluted second was **11**, m.p. 70–72°,  $[\alpha]_{\rm D}^{25}$  –87° (*c* 1, chloroform);  $\nu_{\rm max}^{\rm KBr}$  1757, 1280, 1218, 1140, 1054, 964, 900, and 838 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-n.m.r. data, see Tables I–III. (Found: C, 53.98; H, 6.62; N, 4.03.  $C_{16}H_{23}NO_8$  calc.: C, 53.77; H, 6.48; N, 3.90%.)

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